

Homework VII

due October 27th 2006

Alpha decay of Uranium 238

Uranium 238 ($M_{^{238}\text{U}} = 238.050784 u$) decays in Thorium 234 ($M_{^{234}\text{Th}} = 234.043593 u$) by emitting an α particle of mass $M_{\alpha} = 4.002602 u$. Uranium 238 has a life time $\Delta t = 4.4 \times 10^9$ years. The electric charge of an Uranium nucleus is $+92 e$ while the electric charge of an α particle is $+2 e$ with $e = 1.602 \times 10^{-19} C$.

- 1) What is the kinetic energy K_{α} of the emitted α particle (you can neglect the recoil, $1 u = 931.49401 \text{ MeV} c^{-2}$)
- 2) What is the uncertainty ΔK_{α} of quantum origin on the energy K_{α} ?
- 3) What is the momentum of the α particle? Do you need to use relativistic relations to calculate this momentum ($1 u = 10^{-3} \text{ kg} / N_A$ with $N_A = 6.022142 \times 10^{23}$)? *4e3
- 4) Assuming the α particle is confined in the volume of the nucleus until the decay when it escapes, estimate the size of the nucleus (consider the nucleus to be one-dimensional).
- 5) How much time does it take for the α particle to travel across the nucleus?
- 6) In the classical picture where the α particle is bouncing off the edge of the nucleus until it escapes, how many times does it typically travel across the nucleus before the decay occurs.? The α particle can be considered to have some chance P_{Escape} of escaping each time it approaches the edge of the nucleus. Estimate P_{Escape} .
- 7) Draw a graph of the potential energy for the interaction between the α particle and the rest of the nucleus as a function of the distance from the center of the nucleus. Assume, as we implicitly did, the potential is zero inside the nucleus. Draw a horizontal line across your graph to represent the kinetic energy of the α particle.
- 8) In order for the α particle to escape the nucleus it must violate energy conservation for a short time interval Δt . Estimate Δt from your estimate of the nucleus size and from the repulsive Coulomb potential between the α particle and the rest of the nucleus (hint: estimate the Coulomb potential at $\sim 1 \text{ fm}$ distance, $1/4 \pi \epsilon_0 = 9.987 \times 10^9 \text{ N} \cdot \text{m}^2 \cdot \text{C}^{-2}$).
- 9) How fast must the α particle be traveling when it traverses the Coulomb potential barrier? This result should surprise you. Make a few comments. Can you identify approximations we did that could be changed so we obtain a more acceptable result?