

Name: _____

Social Security #: _____

34 PTS

TA (circle one): Bird Denholm Harrison Johnston Roberts Wilcox

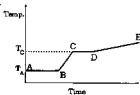
2 A. *60 POINTS*
The drawing to the right shows glass tubing, a rubber bulb and two bottles. Is the situation you see possible? If so, carefully describe what has taken place in order to produce the situation depicted. *THE SITUATION IS POSSIBLE. THE DENSITY OF THE LIQUID IN BOTTLE L < P₂. INITIALLY THE BULB IS SQUEEZED AND RELEASED...*
AS A RESULT THE PRESSURE IN THE GLASS TUBE IS P₀ AND THE LIQUIDS WILL RISE UP THE TUBES. IN BOTH CASES P₀₀ + P₀g h = P₀
THUS, P₁g h_R = P₂g h_L AND SINCE P₂ < P₁, h_L > h_R



IGNORE
150 POINTS

initially at temperature T_A

B. Heat is added to an unknown solid at a uniform rate. The temperature-time plot is shown to the right.



- AB Region where solid is melting.
- CD Region where liquid vaporizes.
- MELTING The process involving the larger latent heat.
- DE (VAPOR) The region showing the larger specific heat capacity.
- T_A The melting temperature of the solid.
- T_C The boiling temperature of the liquid.

3 PTS EACH

C. *10 PTS* The picture depicts three glass vessels, each filled with a liquid. The liquids each have different densities, and $\rho_A > \rho_B > \rho_C$. In vessel B sits an unknown block halfway to the bottom. *and completely submerged.*



A



B



C

- C In which vessel would the block sit on the bottom?
- A In which vessel would the block float on the top?
- C In which vessel would the block feel the smallest buoyant force?
- A & B In which vessels are buoyant forces on the block are the same?
- SINK Assume the coefficient of volume expansion for B and the block are $\beta_B > \beta_{\text{block}}$. If the temperature of vessel B with the block is raised, would block B rise to the surface, sink to the bottom, or remain where it is?

2 PTS EACH

the liquid in