



THIRD MIDTERM

Name (print) C. Gundlach Name (signed) \_\_\_\_\_

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Discussion Section # \_\_\_\_\_

-1 UNITS  
 -1 SF

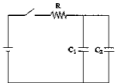
**REPORT ALL NUMBERS TO THREE SIGNIFICANT FIGURES!**  
 Use the conversion constants and data given on the front page.

(a) For the circuit shown, calculate the time constant.

$R = 1.45 \times 10^3 \Omega$   
 $C_1 = 25.0 \text{ pF}$   
 $C_2 = 40.0 \text{ pF}$

C/A or  $\Omega F$  was not counted  
 as a correct unit

$$\tau = R(C_1 + C_2) = 9.43 \cdot 10^{-8} \text{ s} = 94.3 \text{ ns}$$



(b) Calculate the capacitance of a parallel plate capacitor if the plates are square, with sides of 0.250 m, a separation of 0.220 mm, and a dielectric of  $\kappa = 4.50$  filling the space between the plates.

$$C = \kappa A \epsilon_0 / d = \kappa \ell^2 \epsilon_0 / d = 1.131 \cdot 10^{-8} \text{ F} = 11.31 \text{ nF}$$

(c) A copper wire has a diameter of 2.30 mm. If it carries a current of 17.2 A, calculate the magnitude of the current density in  $\text{A/m}^2$ .

$$j = I/A = I/(\pi r^2) = 4.14 \cdot 10^6 \text{ A/m}^2$$

(d) Antimony (a semimetal) has atomic mass of 121.8 and a density of  $6.68 \text{ g/cm}^3$ . If we assume it has  $1.25 \times 10^{24}$  charge carriers per atom, what is the density of charge carries per  $\text{m}^3$ ?

$$(1.25 \cdot 10^{24}) (6.68 \cdot 10^6 \text{ g/cm}^3) (121.8 \text{ g/mol})^{-1} (6.02 \cdot 10^{23} / \text{mol}) = 4.13 \cdot 10^{24} / \text{m}^3$$

(e) All resistances in the network have a resistance R. Calculate the effective resistance between a and b.

$$R_{\text{eff}} = \left\{ \left[ (3R)^{-1} + (2R)^{-1} \right]^{-1} + R^{-1} \right\}^{-1}$$

$$= \left\{ \left[ \frac{1}{3} + \frac{1}{2} \right]^{-1} + 1 \right\}^{-1} R = \frac{5}{11} R = 0.455 R$$

